



## Mathematics and Science | Faculty of Chemistry and Food Chemistry | Physical Chemistry

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### What are they good for?

Due to their unique properties, they can be used in the future to increase the efficiency and long-term stability of fuel cells. Another approach is to use

# Pt\_Pd\_100-n Aerogels for Oxygen Reduction ReactionImage: Chem. Int. Ed. 2017, 56, 13200-132217Image: Chem. Int. Ed. 2017, 56, 13200-132017Image: Chem. Int. Ed. 2017, 56, 13200-132017Image: Chem. Int. Ed. 2017,

Metal aerogels are solid, nanostructured networks that contain a lot of air (up to 95 %). They can be imagined as metal sponges, which are composed of pores and building blocks in the nanometer sized regime.



# Why are they special?

They have extremely large specific surface areas (4-5 soccer fields per mol), low densities and high porosities. them as catalysts, for example for the conversion of  $CO_2$  to usable basic chemicals.



ngew. Chem. Int. Ed. 2015, 54, 13101 –13105

Physical Chemistry (PC) is one of the fundamental areas of chemistry and is part of the mandatory program in every chemistry degree. It combines chemistry and physics, whereby the physical laws are investigated on objects of chemistry.

AEROGELS



# By the way ...what are nanoparticles?

Nanoparticles (NPs) are materials with at least one dimension in the nanoscale (10<sup>-9</sup> m). They have unique chemical and physical properties that can be tuned synthetically by their shape size, and composition. In our group, especially metal semiconductor NPs are in the focus.

Prof. Dr. Alexander Eychmüller is leading the research group of "Physical Chemistry" at TU Dresden since 2005. Our focus is on the development, synthesis and characterization of novel nanoparticles and aerogels with the aim to use them later for various applications, such as catalysts or electronic devices.

# NANOPARTICLES

Synthesis of NPs
\* Zn<sup>2+</sup>, Sn<sup>4+</sup>
Common approaches
are e.g. direct synthesis,
+ Or cation exchange in
\* Pb<sup>2+</sup>
\* Zn<sup>2+</sup> either organic or
\* Sn<sup>4+</sup>
inorganic solvents

 Characterization

 Then, we characterize

 our NPs with

 our NPs with



MoS<sub>2</sub>

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