

A diffuse-interface method for two-phase flows with soluble surfactants

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ABSTRACT

A method is presented to solve two-phase problems involving soluble surfactants. The incompressible Navier–Stokes equations are solved along with equations for the bulk and interfacial surfactant concentrations. A nonlinear equation of state is used to relate the surface tension to the interfacial surfactant concentration. The method is based on the use of a diffuse interface, which allows a simple implementation using standard finite difference or finite element techniques. Here, finite difference methods on a block-structured adaptive grid are used, and the resulting equations are solved using a nonlinear multigrid method. Results are presented for a drop in shear flow in both 2D and 3D, and the effect of solubility is discussed.

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1. Introduction

The presence of surface active agents (surfactants) at fluid interfaces can have a considerable effect on flow dynamics. Surfactants are amphiphilic organic compounds, which can be adsorbed at liquid/gas or liquid/liquid interfaces. The presence of surfactant typically alters the interface dynamics by a reduction in the surface tension of the interface. An inhomogeneous distribution of surfactants produces gradients in surface tension, which again gives rise to tangential forces along the interface. Through this so-called Marangoni effect, surfactants can play an important role in several physical phenomena such as vortex pair interaction (e.g., [1]), fingering (e.g., [2]), drop break-up and coalescence (e.g., [3–5]) and tip-streaming (e.g., [6,4]).

From a numerical point of view, solving the problem of soluble surfactants is highly challenging. A coupled bulk/surface system of equations must be solved on a moving, complex domain, where the domain boundary may stretch, break-up or coalesce with other interfaces. Adsorption of mass to, and desorption of mass from, the interfaces poses another challenge.

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Further, the surface concentration may only be soluble in either the exterior or interior of the domain (e.g., amphiphilic nature of surfactants). The available numerical methods for solving these problems can roughly be divided into two categories: interface tracking and interface capturing methods. Interface tracking methods use either a separate grid for the interface, or a set of interconnected points to mark the interface. For example, boundary integral methods use a surface mesh to track the interface. In the context of surfactants, a boundary integral method for studying the effect of insoluble surfactants on drop deformation was developed in [7]. This method was extended to arbitrary viscosity ratios in [8], to 3D in [9,4] and to soluble surfactants in [10]. See the review [11]. Another tracking method is the front-tracking method (see the review [12]), where a fixed grid is used to compute the flow, while a set of connected marker particles is used to track the interface and any interfacial quantities. A front-tracking method for insoluble surfactants was developed in [13], and this method was extended to handle soluble surfactants in [14,15]. A related front tracking method is the immersed boundary method (see the review [16]), which was recently used to simulate interfacial flows with insoluble surfactants using a surfactant-conserving algorithm [17]. A ghost-cell immersed boundary method was introduced in [18], and was used to study the effects of a diffusion controlled surfactant on a viscous drop injected into a viscous medium [19]. A hybrid level-set/front-tracking approach was used to study the dynamics of capillary waves with insoluble surfactant [20]. Another front-tracking method which combines a finite element methodology with adaptive body-fitted meshes was used to simulate the deformation and break-up of axisymmetric liquid bridges [21] and thin filaments [22] with insoluble surfactants. Very recently, Booty and Siegel [23] developed a hybrid numerical method to simulate bubbles in Stokes flow by combining a boundary integral method with a fixed grid solution of the bulk surfactant equation (using a mapped domain) that incorporates a singular perturbation analysis to account for the rapid variation of the bulk surfactant concentration near the interface when the bulk Peclet number is large. In general, interface tracking methods can be made very accurate but can be relatively complicated to implement, especially in three dimensions and for problems involving topological changes.

In interface capturing methods, the interface is not tracked explicitly, but instead is implicitly defined through an auxiliary function (e.g., level-set, color or phase-field function). This means that the solution of the problem can be done independently of the underlying grid, which greatly simplifies gridding, discretization and handling of topological changes. For example, a volume-of-fluid (VOF) method (see the review [24]) for insoluble surfactants was developed in [25]. A more general method which allows nonlinear equations of state for surface tension was later developed in [26]. Very recently, a VOF method was developed for soluble surfactants in the limit of large sorption rates where there is an analytic relationship between the surface and bulk concentrations at the interface [27]. A level-set method [28] for solving the surfactant equation was presented in [29], and later coupled to an immersed-interface external flow solver in [30]. See the review [31] for fluid dynamics applications. An alternative approach for simulating fluid interfaces with insoluble or soluble surfactant was developed in [32,5,33], using the so-called Arbitrary Lagrangian–Eulerian (ALE) method (see the review [34]) together with a coupled level-set and volume of fluid method. Very recently, surfactant dynamics was simulated using a conservative smoothed particle hydrodynamics algorithm [35].

The diffuse-interface, or phase-field, method represents yet another approach for simulating solutions of equations in complex, evolving domains (see the reviews [36,37]). In this method, which we follow here, the complex domain is represented implicitly by a phase-field function, which is an approximation of the characteristic function of the drop or matrix fluid domain. The domain boundary is replaced by a narrow diffuse interface layer such that the phase-field function rapidly transits from one inside the domain to zero in the exterior of the domain. The boundary of the domain can thus be represented as an isosurface of the phase-field function. The bulk and surface PDEs are then extended on a larger, regular domain with additional terms that approximate the adsorption–desorption flux boundary conditions and source terms for the bulk and surface equations respectively. Standard finite-difference or finite-element methods may be used. Here, we focus on a finite difference approach.

The diffuse interface method, which has a long history in the theory of phase transitions dating back to van der Waals (e.g., [38,36,37]), has been used to simulate multiphase flows in simple geometries including drop coalescence and break-up, electrowetting, and viscoelasticity (e.g., [39–47]). In [48], a diffuse interface model is implemented using a lattice Boltzmann scheme to simulate the effect of surfactant adsorption on droplet dynamics; preliminary simulations were performed. The interaction of multiphase flows with complex boundaries has also been investigated using the diffuse interface method including contact line dynamics and the effect of wetting (e.g., [49–51]). More generally, diffuse interface methods have been developed for solving PDEs on stationary surfaces [52], evolving surfaces [53–56] and for solving PDEs in complex stationary [57–59] and evolving domains with Dirichlet, Neumann and Robin boundary conditions [60]. Extending previous work [58], in [61] it was shown how to solve the coupled bulk/surface problem on general, evolving domains with the diffuse interface method.

Here, we couple the diffuse interface approach from [61] with the solution of the Navier–Stokes equations, and use the method to simulate a drop in shear flow in the presence of a soluble surfactant. The method is very simple compared to previous methods, and can handle advection, diffusion and adsorption/desorption in a straightforward manner. The use of a nonlinear multigrid method and block-structured, adaptive grids also make the method computationally efficient. Results are presented for a drop in shear flow in both 2D and 3D, and the effect of surfactant solubility is discussed. As observed by Milliken and Leal [10] in the context of drop deformation in extensional flows, solubility mitigates many of the surfactant effects by making the surface concentration more uniform. In particular, while an insoluble surfactant can immobilize a surface, solubility may remobilize the surface due to surfactant exchange with the bulk.

The paper is organized as follows. In Section 2, the mathematical formulation of the problem is stated. The governing equations for the interfacial surfactant and the bulk surfactant are introduced, and the interface representation is presented. Section 3 then details the numerical implementation of the mathematical formulation and Section 4 presents some validation test cases of the method. In Section 5, a drop in shear flow is studied in detail and the effect of solubility discussed. Finally, Section 6 provides conclusions and discusses future work.

2. Mathematical formulation

2.1. Governing equations

Consider a domain $\Omega \subset \mathbb{R}^{2,3}$, which contains a closed interface, Γ . The interior of the interface is $\Omega_0 \subset \Omega$, and the exterior is $\Omega_1 \subset \Omega$. See Fig. 1 for an illustration.

We assume that the flow inside the domain Ω is governed by the incompressible Navier–Stokes equations, which in dimensional form are

$$\begin{aligned} \rho^* \left(\frac{\partial \mathbf{u}^*}{\partial t} + (\mathbf{u}^* \cdot \nabla) \mathbf{u}^* \right) &= -\nabla p^* + \nabla \cdot \left(\mu^* (\nabla \mathbf{u}^* + \nabla (\mathbf{u}^*)^T) \right) + \mathbf{F}^*, \\ \nabla \cdot \mathbf{u}^* &= 0. \end{aligned} \quad (1)$$

where ρ is the density, \mathbf{u}^* is the velocity, p^* is the pressure, μ is the dynamic viscosity, and \mathbf{F}^* is the interfacial force, given below in Eq. (5). The density and viscosity can be discontinuous across the interface. We will only consider cases where the densities and viscosities are equal in the two phases in this work, but the discontinuity can in general be handled by interpolating their values using an indicator or characteristic function χ that varies from 0 to 1 across the interface (characterising function of Ω_1), such that

$$\mu^* = \mu_0(1 - \chi) + \mu_1\chi, \quad (2)$$

$$\rho^* = \rho_0(1 - \chi) + \rho_1\chi. \quad (3)$$

In practice, χ may be smoothed so that the transition from 0 to 1 is steep but finite. An example of a smoothed characteristic function is given in Section 2.3. Using a phase-field function c as described in Section 2.3, the governing equations can then be reposed following either the approach of Ding et al. [47] where the volume-averaged velocity is used or Lowengrub and Truskinovsky [38] where the mass-averaged velocity is used (see also [62,63]).

We consider the deformation of an initially circular drop of radius a , placed in a shear flow of shear rate $\dot{\gamma}$. Choosing $\mathcal{L} = a$ as the length scale and $\mathcal{T} = \dot{\gamma}^{-1}$ as the time scale, and $\mathcal{U} = \dot{\gamma}a$ as the velocity scale, the nondimensional Navier–Stokes equations can be written as

$$\begin{aligned} \frac{\partial \mathbf{u}}{\partial t} + (\mathbf{u} \cdot \nabla) \mathbf{u} &= -\nabla p + \frac{1}{\text{Re}} \nabla^2 \mathbf{u} + \frac{1}{\text{ReCa}} \mathbf{F}, \\ \nabla \cdot \mathbf{u} &= 0, \end{aligned} \quad (4)$$

where $\text{Re} = \rho_1 \dot{\gamma} a^2 / \mu_1$ is the Reynolds number, which measures the relative strength of the inertial and viscous forces, and $\text{Ca} = \mu_1 \dot{\gamma} / \sigma_0$ is the Capillary number which measures the relative strength of the viscous and surface tension forces, where σ_0 is the surface tension of a clean drop. The influence of these parameters on a clean interface has been extensively studied

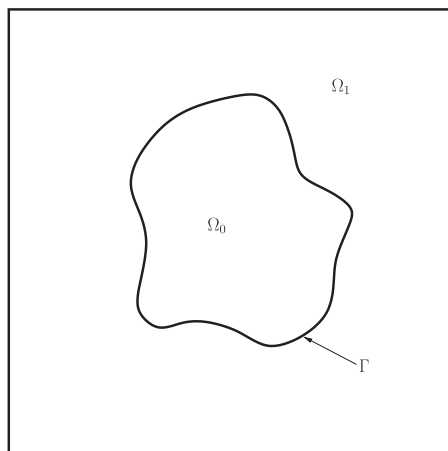


Fig. 1. Illustration of the mathematical domain.

in the literature (e.g., [64–67]). For example, as the capillary number is increased, the drop will become increasingly elongated. There exists a critical capillary number, above which the drop will no longer attain a steady shape, but continue to stretch until it breaks into smaller droplets. The value of the critical capillary number increases with increasing Reynolds numbers and viscosity ratios $\lambda = \mu_0/\mu_1$. Larger viscosity ratios serve to reduce the deformation, and for creeping flow there exists a limit where there is no critical capillary number. However, for flows with higher Reynolds numbers, break-up may still occur, due to inertial forces. In this work, we will only consider flows with moderate Reynolds numbers and viscosity ratios.

The interface force \mathbf{F}^* is given by

$$\mathbf{F}^* = \nabla \cdot ((\mathbf{I} - \mathbf{n} \otimes \mathbf{n})\sigma^* \delta_\Gamma). \quad (5)$$

Here, σ^* is the interfacial tension, \mathbf{I} is the identity tensor and \mathbf{n} is the outward pointing unit normal vector. δ_Γ is a regularized delta function, to be defined in Section 2.3. Alternately, one may write

$$\mathbf{F}^* = -\sigma^* \kappa \delta_\Gamma \mathbf{n} + (\nabla_s \sigma^*) \delta_\Gamma, \quad (6)$$

where κ is the total curvature and $\nabla_s = (\mathbf{I} - \mathbf{n} \otimes \mathbf{n})\nabla$ is the surface gradient operator. The first term on the right hand side is the normal surface tension force, while the second is the tangential (Marangoni) force, which appears due to non-uniform surface tension.

Next, we assume that the interface is covered by surfactants, and we let f^* denote the dimensional surfactant concentration defined on Γ . The presence of a surfactant will decrease the surface tension of the interface. We assume that the dependence of the surface tension is governed by the Langmuir equation of state,

$$\sigma^*(f^*) = \sigma_0 \left[1 + \frac{\mathcal{R}Tf_\infty}{\sigma_0} \ln \left(1 - \frac{f^*}{f_\infty} \right) \right], \quad (7)$$

where \mathcal{R} is the ideal gas constant, T is the absolute temperature, σ_0 is the surface tension of a clean interface and f_∞ is the maximum interfacial surfactant concentration. Let f_e denote the average of f^* at time $t = 0$. We define $f = f^*/f_e$ to be the dimensionless surfactant concentration, which gives the dimensionless surface tension

$$\sigma(f) = 1 + \beta \ln(1 - \chi f), \quad (8)$$

where $\beta = \mathcal{R}Tf_\infty/\sigma_0$ is the elasticity number and $\chi = f_e/f_\infty$ is the dimensionless surfactant coverage. The elasticity number is a measure of the sensitivity of the surface tension to the surfactant concentration, and thus a larger elasticity number increases the deformation. In this definition, β is independent of χ , see [68].

For low surfactant concentrations, Eq. (8) can be simplified to

$$\sigma(f) = 1 - \beta \chi f, \quad (9)$$

which is known as the linear equation of state. These two expressions are compared in Fig. 2. We see that the expressions are similar for low surfactant concentrations, but deviate significantly for higher concentrations. Also note that for large elasticity numbers and surfactant concentrations close to $1/\chi$, the nonlinear equation of state may give unphysical (negative) values of surface tension. In [15], a minimum surface tension was introduced to alleviate this problem. However, in [69], it was ar-

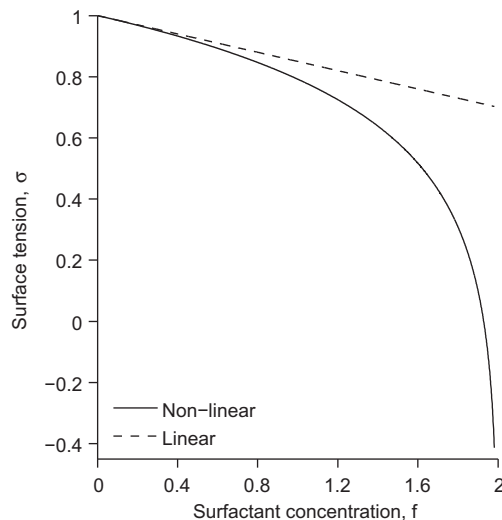


Fig. 2. Dependence of surface tension on surfactant concentration. Comparison of a linear and nonlinear equation of state. $\beta = 0.3$, $\chi = 0.5$.

gued that high elasticity numbers rarely occur in real systems. We use Eq. (8) directly for all the simulations in this work and do not encounter negative surface tension.

In [70], the sharp-interface representation of the surfactant mass balance equation is

$$\frac{df^*}{dt} - \mathbf{u}^* \cdot \nabla_s f^* + f^* (\mathbf{u}^* \cdot \mathbf{n}) (\nabla_s \cdot \mathbf{n}) = -\nabla_s \cdot (f^* \mathbf{u}_s^*) + D_f \nabla_s^2 f^*, \quad (10)$$

where $\frac{d}{dt}$ is the material derivative, $\mathbf{u}_s^* = (\mathbf{I} - \mathbf{n} \otimes \mathbf{n}) \mathbf{u}^*$ is the tangential velocity, D_f is the diffusion coefficient, and j^* is the source term. Assuming that f^* may be extended off Γ , the time derivative term can be changed to

$$\frac{df^*}{dt} = \frac{\partial f^*}{\partial t} + \mathbf{u}^* \cdot \nabla f^*. \quad (11)$$

Since

$$-\mathbf{u}^* \cdot \nabla_s f^* = -\mathbf{u}^* \cdot \nabla f^* + (\mathbf{u}^* \cdot \mathbf{n}) \mathbf{n} \cdot \nabla f^*, \quad (12)$$

$$f^* (\mathbf{u}^* \cdot \mathbf{n}) (\nabla_s \cdot \mathbf{n}) = \kappa f^* \mathbf{u}^* \cdot \mathbf{n}, \quad (13)$$

$$\nabla_s \cdot (f^* \mathbf{u}_s^*) = \mathbf{u}^* \cdot \nabla f^* - (\mathbf{u}^* \cdot \mathbf{n}) \mathbf{u} \cdot \nabla f^* - \kappa f^* \mathbf{u}^* \cdot \mathbf{n} - f^* \mathbf{n} \cdot \nabla \mathbf{u}^* \cdot \mathbf{n}, \quad (14)$$

$$\nabla_s^2 f^* = (\mathbf{I} - \mathbf{n} \otimes \mathbf{n}) \nabla \cdot (\mathbf{I} - \mathbf{n} \otimes \mathbf{n}) \nabla f^* = \nabla^2 f^* - \mathbf{n} \cdot \nabla \nabla f^* \cdot \mathbf{n} - \kappa \mathbf{n} \cdot \nabla f^*, \quad (15)$$

Eq. (10) is equivalent to

$$\frac{\partial f^*}{\partial t} + \mathbf{u}^* \cdot \nabla f^* - f^* (\mathbf{n} \cdot \nabla \mathbf{u}^* \cdot \mathbf{n}) = D_f (\nabla^2 f^* - \mathbf{n} \cdot \nabla \nabla f^* \cdot \mathbf{n} - \kappa \mathbf{n} \cdot \nabla f^*) + j^*. \quad (16)$$

The source term, j^* , is given by

$$j^*(f^*, F^*) = r_a F_s^* (f_\infty - f^*) - r_d f^*, \quad (17)$$

where r_a and r_d are adsorption and desorption coefficients, respectively, and F_s^* is the bulk surfactant concentration immediately adjacent to the surface. For dilute concentrations, this can be simplified to

$$j^*(f^*, F^*) = r_a f_\infty F_s^* - r_d f^*. \quad (18)$$

In this work, we use the full nonlinear form in Eq. (17).

Now, assume that f^* is soluble in Ω_1 , but not in Ω_0 . The bulk concentration, F^* , in Ω_1 , evolves according to

$$\frac{\partial F^*}{\partial t} + \nabla \cdot (F^* \mathbf{u}^*) = D_f \nabla^2 F^* \text{ in } \Omega_1, \quad (19)$$

with the boundary condition at Γ

$$D_f \nabla F^* \cdot \mathbf{n} = -j^* \text{ on } \Gamma. \quad (20)$$

The equation for the surface surfactant concentration, Eq. (16), can be extended to the general domain Ω by introducing a surface delta function, δ_Γ , such that

$$\int_\Gamma f^* d\Gamma = \int_\Omega f^* \delta_\Gamma d\Omega. \quad (21)$$

Using that (e.g., [26])

$$\partial_t \delta_\Gamma + \mathbf{u} \cdot \nabla \delta_\Gamma = -(\mathbf{n} \cdot \nabla \mathbf{u} \cdot \mathbf{n}) \delta_\Gamma,$$

Eq. (16) can be rewritten in distribution form as [61]

$$\frac{\partial}{\partial t} (f^* \delta_\Gamma) + \nabla \cdot (f^* \delta_\Gamma \mathbf{u}^*) = D_f \nabla \cdot (\delta_\Gamma \nabla f^*) + \delta_\Gamma j^*, \quad (22)$$

where it is understood that f^* is extended as a constant in the normal direction off the interface. Note that in the diffuse interface reformulation of Eq. (22) this is indeed the case to leading order in the interface thickness [61]. The formulation given by Eq. (22) is considerably simpler than the sharp-interface formulation.

Similarly, the bulk concentration Eq. (19) can be extended to Ω by using the characteristic function χ

$$\chi = \begin{cases} 1 & \text{in } \Omega_1, \\ 0 & \text{in } \Omega_0. \end{cases} \quad (23)$$

In distribution form, Eq. (19) becomes

$$\frac{\partial}{\partial t}(\chi F^*) + \nabla \cdot (\chi F^* \mathbf{u}^*) = D_F \nabla \cdot (\chi \nabla F^*) - \delta_{\Gamma} j^*, \quad (24)$$

where the boundary condition has been included using the approach from [60,61].

Let $F = F^*/F_e$ be the dimensionless surfactant concentration in the bulk fluid, where F_e average of the initial bulk concentration. Then, Eqs. (22) and (24) can be written in dimensionless form as

$$\frac{\partial}{\partial t}(f \delta_{\Gamma}) + \nabla \cdot (f \delta_{\Gamma} \mathbf{u}) = \frac{1}{\text{Pe}_f} \nabla \cdot (\delta_{\Gamma} \nabla f) + \delta_{\Gamma} j, \quad (25)$$

$$\frac{\partial}{\partial t}(\chi F) + \nabla \cdot (\chi F \mathbf{u}) = \frac{1}{\text{Pe}_F} \nabla \cdot (\chi \nabla F) - h \delta_{\Gamma} j, \quad (26)$$

where $\text{Pe}_f = \dot{\gamma} a^2 / D_f$ is the surface Peclet number, $\text{Pe}_F = \dot{\gamma} a^2 / D_F$ is the bulk Peclet number and $h = f_e / (a F_e)$ is the adsorption depth. The dimensionless source term, j , is given by

$$j(F, f) = \text{Bi} \left[k F_s \left(\frac{1}{x} - f \right) - f \right]. \quad (27)$$

The dimensionless parameters are $\text{Bi} = r_d / \dot{\gamma}$ (Biot number) and $k = r_d F_e / r_a$ (adsorption number). In our simulations, we assume that f_e and F_e are in equilibrium such that $j^*(f_e, F_e) = 0$. Consequently, the following relations hold:

$$k = \frac{x}{1-x}, \quad (28)$$

$$h = \frac{r_d f_{\infty}}{r_d a} (1-x). \quad (29)$$

An insoluble surfactant may be characterized by $\text{Bi} = 0$. The Biot number is the ratio of the desorption rate to the interfacial surfactant convection due to shear. For large Biot numbers, there is strong coupling between the surface and the bulk.

2.2. Interface representation

To represent the interface implicitly, we may either use a level-set function or a phase-field function. For example, taking $r(\mathbf{x}, t)$ to be a signed distance function from Γ to \mathbf{x} , the interface Γ is the $r = 0$ isosurface. Alternatively, using a phase field function c , which is 1 in Ω_0 and 0 in Ω_1 , we may define $\Gamma(t) = \{\mathbf{x} \in \Omega | c(\mathbf{x}, t) = \frac{1}{2}\}$. We follow the latter here and evolve the phase field function using an advective Cahn–Hilliard equation

$$\frac{\partial c}{\partial t} + \nabla \cdot (c \mathbf{u}) = \nabla \cdot (M(c) \nabla \mu_c), \quad (30)$$

$$\mu_c = g'(c) - \varepsilon^2 \nabla^2 c, \quad (31)$$

where $M = \sqrt{c^2(1-c)^2}$ is a mobility function, and $g = c^2(1-c)^2/4$ is a double well potential. The function μ_c is called the chemical potential and ε is a small parameter related to the interface thickness. Note that the above equation system is fourth-order and nonlinear, which requires specialized numerical methods to solve in an efficient manner. Note that for small ε , the following relation holds near Γ

$$c(\mathbf{x}, t) \approx \frac{1}{2} \left[1 - \tanh \left(\frac{r(\mathbf{x}, t)}{2\sqrt{2}\varepsilon} \right) \right]. \quad (32)$$

2.3. Regularized delta and characteristic functions

To be able to evaluate Eqs. (3), (25) and (26) numerically, regularizations of the surface delta function and characteristic function are needed. In the phase-field context, several definitions of the delta function are available from the literature. In this work, the approximation from [52],

$$\delta_{\Gamma} \approx B(c) = \frac{3\sqrt{2}}{\varepsilon} c^2(1-c)^2, \quad (33)$$

is used for the surface equation. For the surface tension term and the boundary condition in the bulk equation, the approximation

$$\delta_{\Gamma} \approx |\nabla c| \quad (34)$$

is used. This is to avoid any scaling of the equations. Note that in the surface Eq. (25), the constants in Eq. (33) cancel out in the discretized equations.

The regularized characteristic function, is simply taken as [60]

$$\chi \approx 1 - c. \tag{35}$$

In [61], it was shown that the regularized equations for the interfacial surfactant and the bulk surfactant converge to the sharp interface equations in the sharp interface limit.

3. Numerical methods

This section briefly describes the numerical methods used to solve the above equations. The algorithm follows that developed in [71]. In particular, the equations are discretized using finite differences in space and a semi-implicit time discretization. All variables are defined at cell-centers. A block-structured, adaptive grid is used to increase the resolution around the interface in an efficient manner. The nonlinear equations at the implicit time level are solved using a nonlinear Adaptive Full Approximation Scheme (AFAS) multigrid algorithm.

Special care has to be taken for the temporal discretization. The Cahn–Hilliard system is fourth order in space, and requires the use of an implicit method to avoid severe limitations in the time step. Here, we employ a combination of explicit Adams–Bashforth schemes for the convective terms and implicit Crank–Nicholson schemes for other terms.

First, the Cahn–Hilliard system is solved to find the phase-field function at the new time step. This is solved using

$$\frac{c^{k+1} - c^k}{\Delta t} = -\nabla \cdot (\mathbf{u}c)^{k+\frac{1}{2}} + \frac{1}{2} \left(\nabla_d \cdot (M^{k+1} \nabla_d \mu_c^{k+1}) + \nabla_d \cdot (M^k \nabla_d \mu_c^k) \right), \tag{36}$$

$$\mu_c^{k+1} = g'(c^{k+1}) - \varepsilon^2 \nabla_d^2 c^{k+1}. \tag{37}$$

The operator ∇_d represents the standard, second-order, finite-difference discretization. The convective terms of the form, $\nabla \cdot (\mathbf{u}\phi)$, are discretized using the third-order WENO reconstruction method [72,73]. The WENO reconstruction method has the advantage that it handles steep gradients well, which may occur in the type of dynamics described in this work. Additionally, fewer grid points are needed to achieve a high order solution. This is particularly important for the efficiency of the adaptive grid, because fewer ghost cell values have to be calculated at the boundaries of each grid block.

The velocities and concentration at the half-step are found by extrapolation, $\mathbf{u}^{k+\frac{1}{2}} = \frac{1}{2}(3\mathbf{u}^k - \mathbf{u}^{k-1})$ and $c^{k+\frac{1}{2}} = \frac{1}{2}(3c^k - c^{k-1})$. The velocities at the cell-edges are needed to construct the convective term. These are found by

$$u_{i+\frac{1}{2}j} = \frac{u_{ij} + u_{i+1j}}{2} - \frac{\Psi_{i+1j} - \Psi_{ij}}{h}, \tag{38}$$

$$v_{ij+\frac{1}{2}} = \frac{v_{ij} + v_{ij+1}}{2} - \frac{\Psi_{ij+1} - \Psi_{ij}}{h}, \tag{39}$$

where Ψ is the MAC projection found by solving

$$\nabla_d^2 \Psi = \nabla_d^c \cdot \mathbf{u}^{k+\frac{1}{2}}, \tag{40}$$

with $\nabla \Psi \cdot \mathbf{n} = 0$ on $\partial\Omega$ (assumed to be along Cartesian directions). In the AFAS nonlinear multigrid method used to solve Eqs. (36), (37), the nonlinear term in the chemical potential equation, $g'(c^{k+1})$, is linearized in the smoother (local linearization) by

$$g'(c^m) \approx g'(c^{m-1}) + g''(c^{m-1})(c^m - c^{m-1}), \tag{41}$$

where m is the V-cycle iteration index. See [71] for details.

Next, the surfactant concentration on the surface and in the bulk at the new time step is found. Because of the regularized delta functions and characteristic functions, no special treatment of these equations are needed. We use the method from [61], and solve

$$\frac{(Bf)^{k+1} - (Bf)^k}{\Delta t} = -\nabla \cdot (\mathbf{u}Bf)^{k+\frac{1}{2}} + \frac{1}{2Pe_f} \left(\nabla_d \cdot (B^{k+1} \nabla_d f^{k+1}) + \nabla_d \cdot (B^k \nabla_d f^k) \right) + \frac{1}{2} ((Bj)^{k+1} + (Bj)^k), \tag{42}$$

$$\frac{(\chi F)^{k+1} - (\chi F)^k}{\Delta t} = -\nabla \cdot (\mathbf{u}\chi F)^{k+\frac{1}{2}} + \frac{1}{2Pe_f} \left(\nabla_d \cdot (\chi^{k+1} \nabla_d F^{k+1}) + \nabla_d \cdot (\chi^k \nabla_d F^k) \right) - \frac{1}{2} \left(|\nabla_d c^{k+1}| (hj)^{k+1} + |\nabla_d c^k| (hj)^k \right). \tag{43}$$

as a coupled system, where we use $\chi = 1 - c$. The source terms at the new time step are lagged in the V-cycle of the multigrid algorithm [61], i.e. $j^{k+1} \approx j^{k+1,m-1}$ where m is the V-cycle iteration index. This did not degrade the multigrid performance, the residual is typically reduced by an order of magnitude for each iteration.

The Navier–Stokes equations are solved using a second-order projection method. The details can be found in [74,75,45].

3.1. Mesh adaptivity

We use block-structured, adaptive mesh refinement to increase the resolution around the interface in an efficient manner. Near the interface, patches of overlapping, uniform, Cartesian grids are applied.

At every time step, we check grid cells for refinement using a simple undivided gradient test. Since it is essential for our problem to have a fine resolution in the diffuse interface region, this test marks grid cells where the finite difference of the phase-field function is large. In particular, we mark a cell for refinement if

$$\sqrt{(c_{i+1j} - c_{i-1j})^2 + (c_{ij+1} - c_{ij-1})^2} > C_k, \quad (44)$$

where C_k is the tolerance for level k . We note that more sophisticated refinement criteria could also be applied. For instance, [76] used a criteria based on curvature and vorticity with a level-set method. It may also be important to refine where finite differences of f and F are large; in particular this is needed when Pe_f and/or Pe_F are large. After the cells are marked for refinement, they are grouped together into rectangular patches, and populated with data from the old grid. For a detailed discussion of all the aspects of the adaptive algorithm, the reader is referred to [71].

4. Code validation

In this section, three test cases are presented to validate the proposed numerical method. Because the bulk/surface coupling was thoroughly tested in [61], we will here focus on validating the flow solver and the coupling between the flow solver and the surfactant concentration.

In the numerical implementation, surface quantities are defined at grid points near the interface. To find values on the interface, we use a marching squares algorithm (see e.g., [77]) to generate a set of points on the 0.5 isocontour of the phase-field function. Bilinear interpolation is then used to interpolate the grid values of the surface concentration to these interface points.

4.1. Oscillations of a capillary wave

First a two-phase system without surfactants is considered to test the Navier–Stokes solver and the adaptive grid algorithm. The test problem is the damping of a sinusoidal, capillary wave. For the case of small amplitudes and matched viscosities, an analytical solution for the amplitude was found in [78].

The frequency of a wave with wavenumber $k = 2\pi/\lambda$, where λ is the wave length, is given by

$$\omega_0^2 = \frac{\sigma k^3}{\rho_0 + \rho_1}. \quad (45)$$

Assuming the amplitude is small, the analytical solution for the amplitude $a(t)$ is given in dimensional form by

$$a(t) = \frac{4(1-4b)k^4 v^2}{8(1-4b)k^4 v^2 + \omega_0^2} a_0 \operatorname{erfc}(vk^2 t)^{1/2} + \sum_{i=4}^4 \frac{z_i}{Z_i} \left(\frac{\omega_0^2 a_0}{z_i^2 - vk^2} - \dot{a}_0 \right) \exp[(z_i^2 - vk^2)t] \operatorname{erfc}(z_i t^{1/2}), \quad (46)$$

where the z_i 's are the four roots of the algebraic equation

$$z^4 - 4b(k^2 v)^{1/2} z^3 + 2(1-6b)k^2 v z^2 + 4(1-3b)(k^2 v)^{3/2} z + (1-4b)k^4 v^2 + \omega_0^2 = 0 \quad (47)$$

and $Z_i = (z_2 - z_1)(z_3 - z_1)(z_4 - z_1)$ with Z_2, Z_3, Z_4 obtained by circular permutation of the indices. Further, b is given by

$$b = \frac{\rho_0 \rho_1}{(\rho_0 + \rho_1)^2}. \quad (48)$$

Here, we will consider a computational domain given by a unit square box, with an interface given by

$$y = 0.5 - 0.01 \cos(2\pi x). \quad (49)$$

We choose dimensional variables $\rho_0 = \rho_1 = 18.3$, $\mu_0 = \mu_1 = 0.078$ and $\sigma = 1.0$, giving $\omega_0^2 = 6.778$. Periodic boundary conditions are used in the horizontal direction, no-slip boundary conditions are imposed on the upper and lower boundaries, and an interface thickness parameter of $\varepsilon = 5 \times 10^{-3}$ is used along with a time step of $\Delta t = 5 \times 10^{-3}$.

The initial condition is shown in Fig. 3 (left). An adaptive grid with 16×16 grid points at the root level and three levels of mesh refinement are used ($h_{\max} = 6.25 \times 10^{-2}$ and $h_{\min} = 7.8125 \times 10^{-3}$). There are 8–10 grid points across the interface layer. Fig. 3 (right) shows the evolution of the amplitude compared to the analytical solution. The numerical method is clearly capable of accurately simulating the oscillatory behavior.

4.2. Rising drop in a linear surfactant gradient

In [79], an approximation for the rise velocity of a viscous drop in a linear temperature gradient was found. This approximation is used here to test the implementation of the Marangoni stresses. Instead of a temperature gradient, the interfacial surfactant concentration is assumed to vary linearly in the vertical direction.

In dimensional variables, consider an axisymmetric drop of radius R in a channel of radius $5R$ and height $L = 15R$. Let the interfacial surface concentration be given by

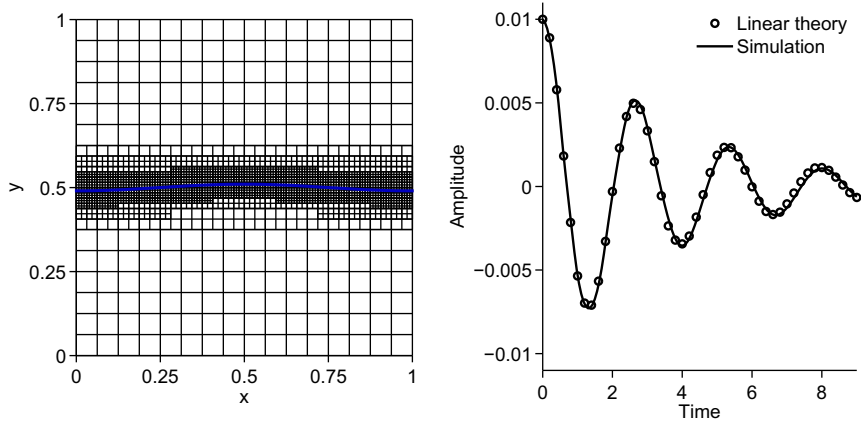


Fig. 3. Capillary wave test case. An adaptive grid with root level 16×16 and three levels of refinement are used, $\varepsilon = 5 \times 10^{-3}$. Left: Computational domain with initial interface position and adaptive grid; Right: Amplitude of capillary wave, simulation versus linear theory.

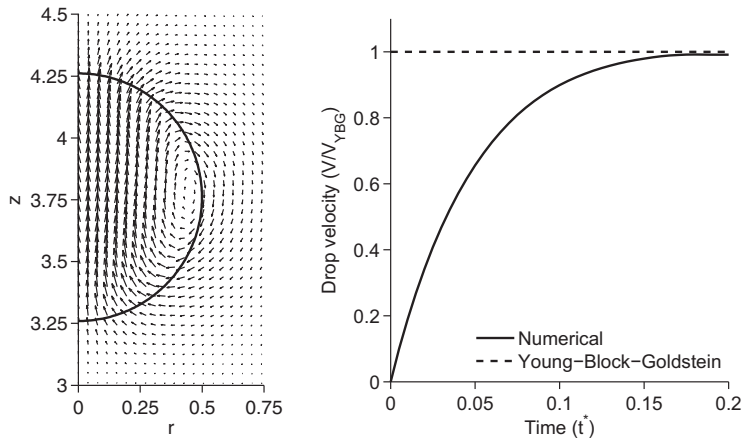


Fig. 4. Velocity field at steady state shown every fourth grid point (left) and rise velocity versus time (right), compared to the theoretical result, for the linear surfactant gradient test case.

$$\frac{f(z)}{f_\infty} = \frac{z}{L}, \tag{50}$$

and the relationship between surfactant and surface tension as

$$\sigma(z) = \sigma_s \left(1 - \beta \frac{f(z)}{f_\infty} \right), \tag{51}$$

then an approximation for the terminal rise velocity is [79]

$$V_{YBG} = \frac{2(\sigma_s \beta_s R / L - \Delta \rho g R^2 (\mu_0 + \mu_1)) / \mu_1}{(6\mu_0 + 9\mu_1)}. \tag{52}$$

We choose dimensional parameters $\rho_0 = \rho_1 = 0.2$, $\mu_0 = \mu_1 = 0.1$, $R = 0.5$, $\sigma_s = 1.0$ and $\beta_s = 1.0$. According to Eq. (52), this should give a Reynolds number of $Re = 0.0889$, which is well within the creeping flow regime for which the equation is valid.

Because of the relatively large domain needed to keep the boundary conditions from interfering with the solution, this test case lends itself very well to an adaptive grid. Here, a root level grid spacing of $h_{\max} = 5/64 = 7.8125 \times 10^{-2}$ is chosen with three levels of refinement $h_{\min} = 5/512 = 9.76 \times 10^{-3}$. The number of nodes is approximately 9000 throughout the simulation, compared to the 196608 nodes needed for a uniform grid. An interface thickness parameter of $\varepsilon = 6.0 \times 10^{-3}$ is used (8–10 grid points across the interface), and the time step is $\Delta t = 5.0 \times 10^{-4}$. The rise velocity is calculated by

$$V = \frac{\int_{\Omega} c(\mathbf{x}, t) \mathbf{u}(\mathbf{x}, t) \cdot \mathbf{e}_z d\Omega}{\int_{\Omega} c(\mathbf{x}, t) d\Omega}, \tag{53}$$

where \mathbf{e}_z is the unit vector in the z -direction. This integral was evaluated using the midpoint rule.

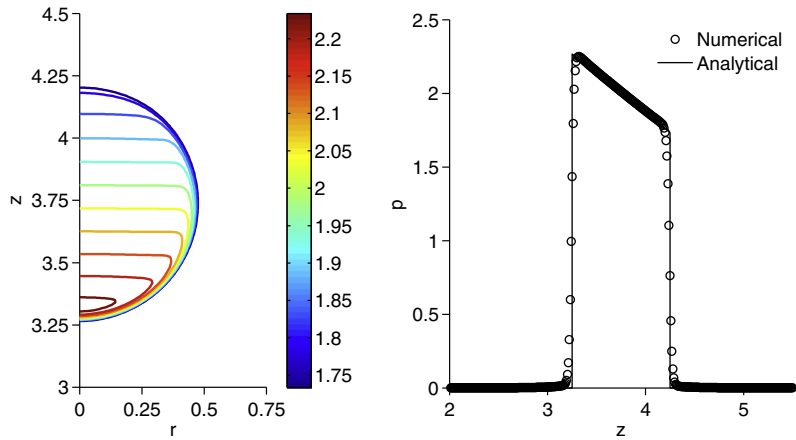


Fig. 5. Pressure after one time step for a drop rising in a linear surfactant gradient. Left: Pressure contours, Right: Pressure along vertical centerline compared to analytical result, Eq. (54). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

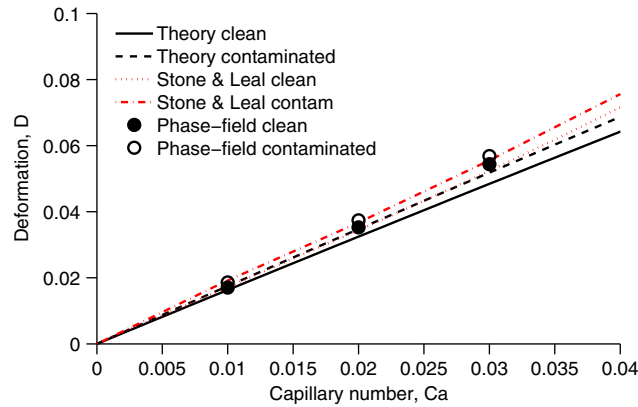


Fig. 6. Comparison of small deformation theory and numerical simulations for a drop stretching in linear flow.

The velocity field around the drop at steady state is shown in Fig. 4 (left). The figure is in good agreement with the corresponding Fig. 10a in [15]. Fig. 4 (right) shows the normalized rise velocity. After an initial acceleration phase, the velocity approaches the theoretical prediction asymptotically.

Assuming the pressure in the surrounding fluid is zero, the pressure inside the drop is given by

$$p(z) = \frac{2\sigma(z)}{R}. \tag{54}$$

Fig. 5 shows a closeup of pressure contours inside the drop (left), and the pressure along the vertical center line compared to the analytical result (right). Good agreement is observed.

4.3. Drop stretching in linear flow

In [7], an expression was presented for the deformation of a surfactant-covered drop in the creeping flow limit under the assumptions $Ca \ll 1$, $Pe_f \ll 1$ and $\lambda = O(1)$. This can be written as

$$D = \frac{L - B}{L + B} \approx \frac{3Ca \cdot b_r}{4 + Ca \cdot b_r}, \tag{55}$$

where L and B denote the drop extension along the z and x axes, respectively, and the coefficient b_r is given by

$$b_r = \frac{(80 + 95\lambda) + \frac{4\beta Pe_f}{Ca(1-\beta)}}{40(1 + \lambda) + \frac{2\beta Pe_f}{Ca(1-\beta)}}. \tag{56}$$

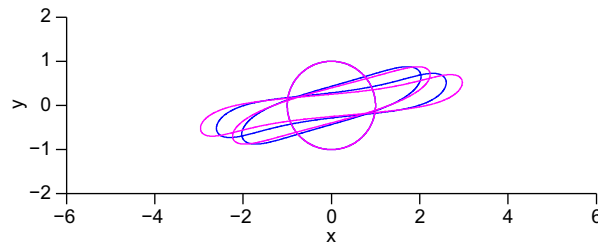


Fig. 7. Comparison of drop shapes for clean and insoluble surfactant-covered drops in a shear flow at times $t = 0, 4, 8$. Blue: Clean, Magenta: Insoluble-surfactant-covered. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

Here, we simulate a 3D drop in a velocity field given by $\mathbf{u} = (-x, -y, 2z)$, with varying Ca . The viscosity ratio is set to $\lambda = 1$ and the surfactant parameters are $Pe = 0.1$ and $\beta = 0.5$. Due to symmetry, we only simulate one octant of the drop, and the domain size is $4R \times 4R \times 4R$. We use an adaptive mesh with three levels of adaptivity, where the finest level has grid spacing $h = 1/64$. The interface thickness parameter is $\varepsilon = 0.025$.

Fig. 6 shows the simulated results together with small deformation theory and results from [7]. We see that there is good agreement with the theory at low Ca and then the discrepancy gets larger at higher Ca . Our results are also in good agreement with the boundary-integral simulations of [7].

5. Influence of soluble surfactant on drop deformation and break up in shear flow

In this section, we consider the deformation of an initially circular drop placed in a shear flow. We begin by presenting 2D results. The (nondimensional) computational domain chosen for the simulations is a rectangular domain of size 12×4 . The nondimensional velocity, $u = y$ is imposed on the upper and lower boundaries, and periodicity is assumed in the horizontal direction. In effect, we are simulating an infinite array of drops placed 8 units (in drop radii) apart. The root level grid spacing is $h_{\max} = 1/16$, and three levels of refinement are used ($h_{\min} = 1/128$), unless otherwise specified. The interface thickness parameter is $\varepsilon = 1/160$, and we choose a time step of 1×10^{-3} . There are approximately 8–10 grid points across the interface layer. We use $Ca = 0.5$, $Re = 1.0$, viscosity ratio $\lambda = 1.0$, and $Pe_f = 10$ for all simulations unless otherwise noted.

5.1. Insoluble surfactant

First, we compare the evolution of a clean drop with a drop covered with insoluble surfactant using $Bi = 0$. We take $\beta = 0.3$, $\chi = 0.5$ and the initial surfactant distribution to be uniform: $f|_{t=0} = 1.0$.

Fig. 7 shows the morphology of a clean drop and an insoluble surfactant-covered drop at times 0, 4 and 8. As expected, the surfactant-covered drop exhibits a larger deformation and rotates faster due to the lower surface tension. Fig. 8 shows the surface quantities surfactant concentration, surface tension, capillary force and Marangoni force as a function of arc length s at times 0, 4 and 8. The point $s = 0$ corresponds to the point $y = 0$ on the right side of the drop and s increases counterclockwise along the drop. The capillary forces are calculated as $-\frac{1}{ca}\sigma\kappa$, while the Marangoni forces are calculated as $\frac{1}{ca}\nabla_s\sigma \cdot \mathbf{t}$, where \mathbf{t} is the unit tangential vector. Surfactant is swept to the drop tips, which lowers the surface tension. This reduces the drop the resistance to stretching. Correspondingly, the drop thins, elongates and rotates. The capillary forces reach a maximum at the tips due to the high curvature. The presence of surfactant does not alter the capillary forces significantly, but does lower the forces near the tips (not shown). Due to symmetry, the Marangoni forces are zero at the drop tips. Near the tips, the Marangoni force has local maxima and minima. The maxima occur on the upper part of the drop interface near the right tip and on the lower part of the interface near the left tip. The minima occur on the lower region near the right tip and the upper region near the left tip. The velocity near the maxima is larger, which leads to increased surfactant convection and thereby larger surface tension gradients.

A common problem with interface capturing methods is that they do not exactly conserve mass. In the above simulation, there is about a 0.3 % loss of surfactant mass throughout the simulation. The mass is measured by evaluating $m_f = \int_{\Omega} Bf d\Omega$ using the midpoint rule. In [30], a simple mass correction scheme was used where the surfactant is multiplied by a constant to preserve mass. This scheme could also be easily adapted to our approach. The mass loss for the phase-field concentration was only $5.6 \times 10^{-5}\%$ (measured by the total integral $\int_{\Omega} c d\Omega$ using the midpoint rule).

5.2. Soluble surfactant

We next investigate the effect of surfactant solubility. We again take $\beta = 0.3$, $\chi = 0.5$, and the initial surface surfactant distribution to be uniform: $f|_{t=0} = 1.0$. The adsorption number is $k = 1.0$ and the adsorption depth is $h = 0.5$. We take the drop to be in equilibrium with the bulk at the start of the simulation, i.e. $j|_{t=0} = 0$, which gives the initial bulk concentration $F|_{t=0} = 1.0$. We vary the Biot number and the bulk Peclet number.

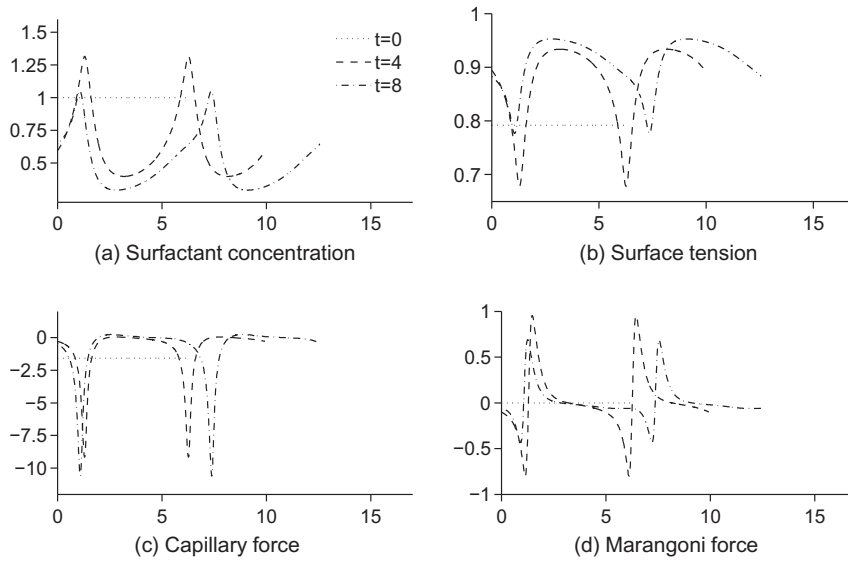


Fig. 8. Surface quantities as a function of arclength for an insoluble surfactant-covered drop in shear flow.

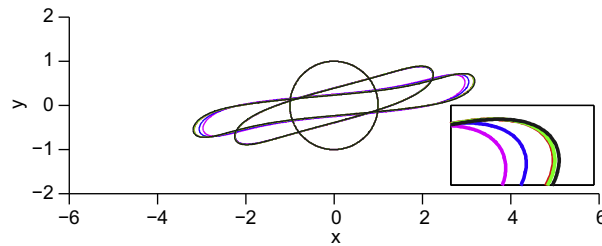


Fig. 9. Comparison of drop shapes for various Biot numbers. The times shown are $t = 0, 4, 8$. Magenta: $Bi = 0$ (insoluble), blue: $Bi = 0.1$, red: $Bi = 1$, green: $Bi = 10$, and black: uniform ($Bi = \infty$). The inset shows a close-up of the drop tip for $t = 8$. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

5.2.1. Influence of Biot number

We first set the bulk Peclet number $Pe_F = 1$, and vary the Biot number. The morphology of the drops for varying Biot numbers is shown in Fig. 9. The inner (magenta) drop corresponds to $Bi = 0$ and the outer drop (black) corresponds to $Bi = \infty$, simulated by enforcing $f = 1$ throughout the simulation. The drop deformation is an increasing function of Bi . The reason for this is that surfactant adsorption/desorption decreases the surface tension gradients and hence Marangoni forces. If the Biot number tends to infinity, the surface surfactant should be uniform and the Marangoni force vanishes.

This is further elucidated in Fig. 10. The left column of Fig. 10 shows the surfactant concentration, the surface tension, the capillary force and the Marangoni force at $t = 4$. We see that by increasing the Biot numbers, the surfactant concentration becomes more uniform. This leads to lower surface tension gradients and thereby lower Marangoni forces, which enables the drop to deform more. The capillary force is dominated by the curvature, and remains largely unaffected by the variation in surface tension.

The right column of Fig. 10 shows the bulk concentration at $t = 4$. It is evident that a larger Biot number leads to a faster depletion of bulk surfactant due to the higher adsorption at the interface. At the drop tips, there is desorption of surfactant off the interface leading to increased surfactant concentration in the bulk. Also note that the change in dynamics from $Bi = 1$ to $Bi = 10$ is not as significant as the change from $Bi = 0.1$ to $Bi = 1$. This is because the bulk diffusion is not large enough to maintain a high adsorption/desorption rate. The process has become *diffusion-limited*.

5.2.2. Influence of Peclet number

Next, we examine the effect of the bulk Peclet number Pe_F . The Biot number is $Bi = 1$. Fig. 11 shows the morphology of the drops for varying Peclet numbers. The deformation is larger for smaller Peclet numbers since stronger diffusion in the bulk enables the redistribution of bulk surfactant to maintain a more uniform bulk distribution. This in turn supports a larger rate of surfactant adsorption along the drop sides that keeps the surface surfactant concentration more uniform and thus decreases the Marangoni force.

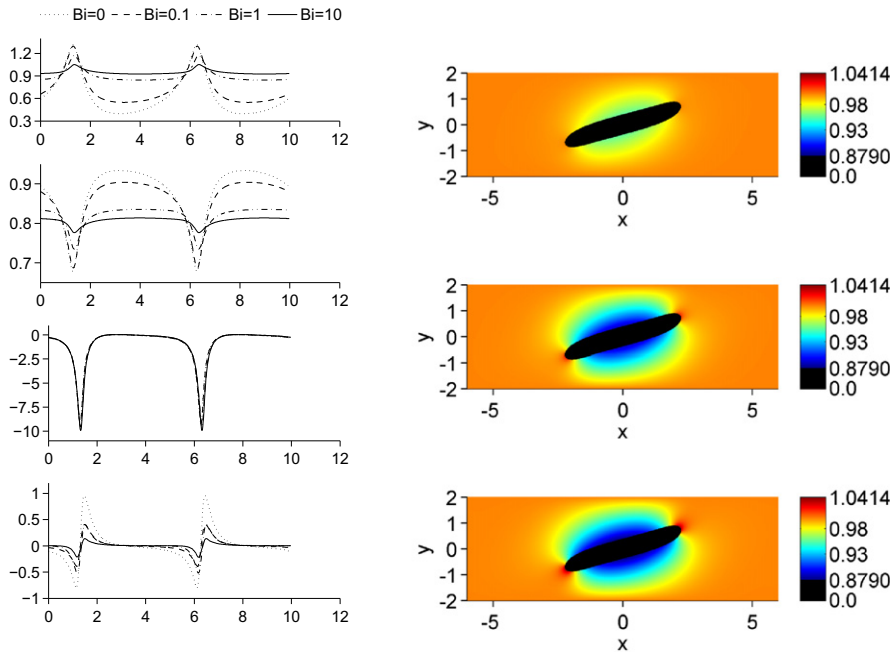


Fig. 10. Influence of the Biot number on a drop in shear flow at $t = 4$. Left column shows from top to bottom: Surfactant concentration, surface tension, capillary force and Marangoni force. The right column shows the bulk surfactant concentration, from top to bottom: $Bi = 0.1$, $Bi = 1$ and $Bi = 10$.

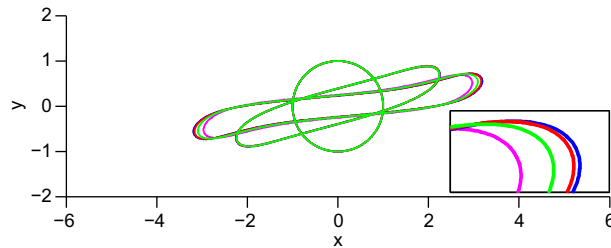


Fig. 11. Comparison of drop shapes for various Peclet numbers. The times shown are $t = 0, 4, 8$. Magenta: Insoluble, blue: $Pe_F = 0.1$, red: $Pe_F = 1$ and green: $Pe_F = 10$. The inset shows a close-up of the drop tip for $t = 8$. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

The left column of Fig. 12 shows the surfactant concentration, the surface tension, the capillary force and the Marangoni force at $t = 4$. Again, we see that the surface surfactant concentration becomes more uniform due to the mass transfer process when the Peclet number is smaller, thereby decreasing the Marangoni forces. However, the effect is most pronounced away from the drop tips. At the drop tips, the surfactant concentration (surface tension and Capillary force) are somewhat insensitive to Pe_F because convection is relatively more important here than on the sides. In particular, at the sides surfactant is adsorbed to the interface, then swept by convection towards the tips, leaving room for more adsorption along the sides. This depletes the bulk concentration near the sides but at the tips, there is increased bulk concentration above $F_\infty = 1$ because surfactant desorbs to the bulk. A small Peclet number is needed to redistribute bulk surfactant to maintain a more uniform bulk distribution and thus a high adsorption rate along the drop sides. In fact, when the Peclet number is small, the adsorption rate is not large enough to adsorb all the available bulk surfactant. Thus the differences between the results from $Pe_F = 0.1$ to $Pe_F = 1$ are less pronounced than the differences from $Pe_F = 1$ to $Pe_F = 10$. This is because the process becomes *adsorption-limited* for small Peclet numbers.

This is further illustrated in Fig. 14, which shows the source term j for the three Peclet numbers. Away from the drop tips, j is positive, which means that surfactant is transported from the bulk to the interface. At the tips, the transportation is reversed due to the accumulation of surfactant here. For a high Peclet number, the mass transfer is lower, because the bulk diffusion is not large enough to transport desorbed surfactant away from the interface or to replenish surfactant adsorbed to the interface. This effect is even more pronounced when the Peclet number is significantly increased to $Pe_F = 1000$ as seen in Fig. 13. At this large Peclet number, a narrow boundary layer of depleted bulk surfactant is observed along the drop sides

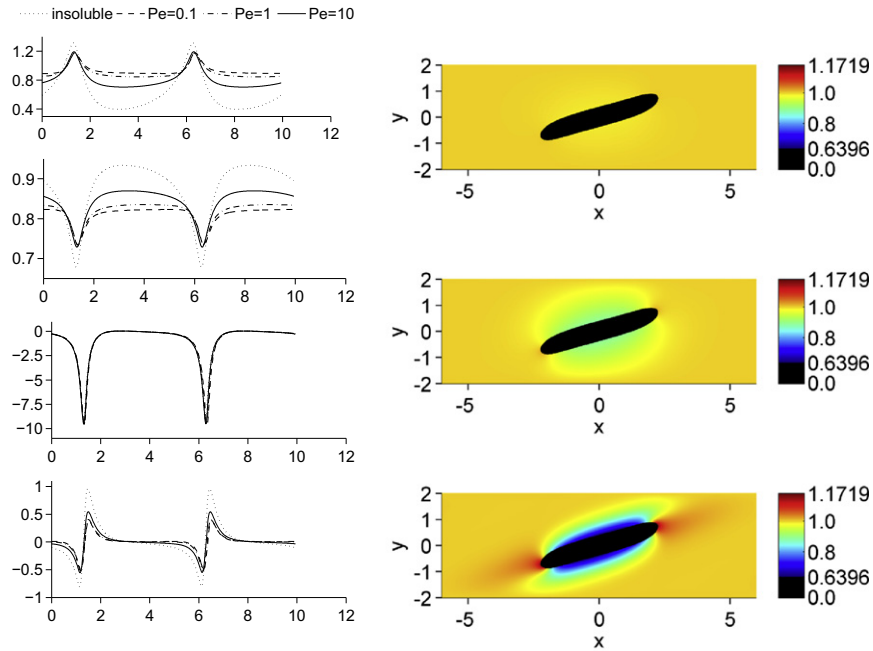


Fig. 12. Influence of bulk Peclet number on drop in shear flow at $t = 4$. Left column shows from top to bottom: Surfactant concentration, surface tension, capillary force and Marangoni force. The right column shows the bulk surfactant concentration, from top to bottom: $Pe_F = 0.1$, $Pe_F = 1$ and $Pe_F = 10$. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

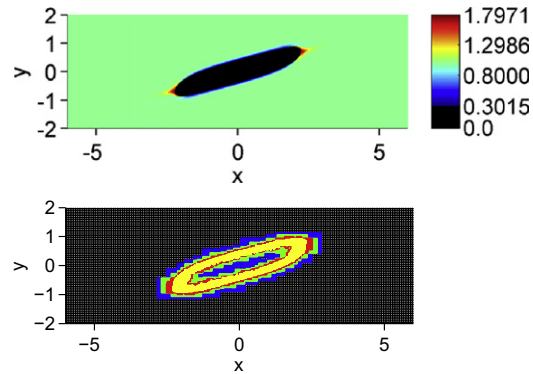


Fig. 13. Influence of large bulk Peclet number $Pe_F = 1000$ on a drop in shear flow at $t = 4$. Top: Bulk surfactant distribution, Bottom: Adaptive mesh. Four levels of refinement are used (black: level 0, blue: level 1, green: level 2, red: level 3, yellow: level 4). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

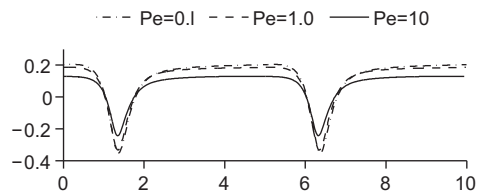


Fig. 14. Comparison of the source term j along the interface for different Peclet numbers.

while the bulk concentration is significantly increased near the drop tips. In order to simulate this case, another level of refinement is required such that $h_{\min} = 1/256$ and in addition the mesh is refined where the undivided gradient of F is sufficiently large; the finest mesh covers the surfactant boundary layer. The thickness of the surfactant boundary layer along the

Table 1
Error in L_2 norm and convergence rate for the fluid velocities and the surfactant concentrations.

Grid	$\ u - u_{\text{ref}}\ _2 (\times 10^{-2})$	Rate	$\ v - v_{\text{ref}}\ _2 (\times 10^{-1})$	Rate
1/16	4.03	-	8.49	-
1/32	1.84	1.13	5.20	0.71
1/64	0.503	1.87	1.57	1.73
1/128	0.0754	2.74	0.203	2.95
	$\ f - f_{\text{ref}}\ _2 (\times 10^{-1})$	Rate	$\ F - F_{\text{ref}}\ _2 (\times 10^{-1})$	Rate
1/16	2.93	-	5.96	-
1/32	2.23	0.39	3.64	0.71
1/64	1.58	0.50	1.19	0.92
1/128	0.686	1.20	0.834	1.20

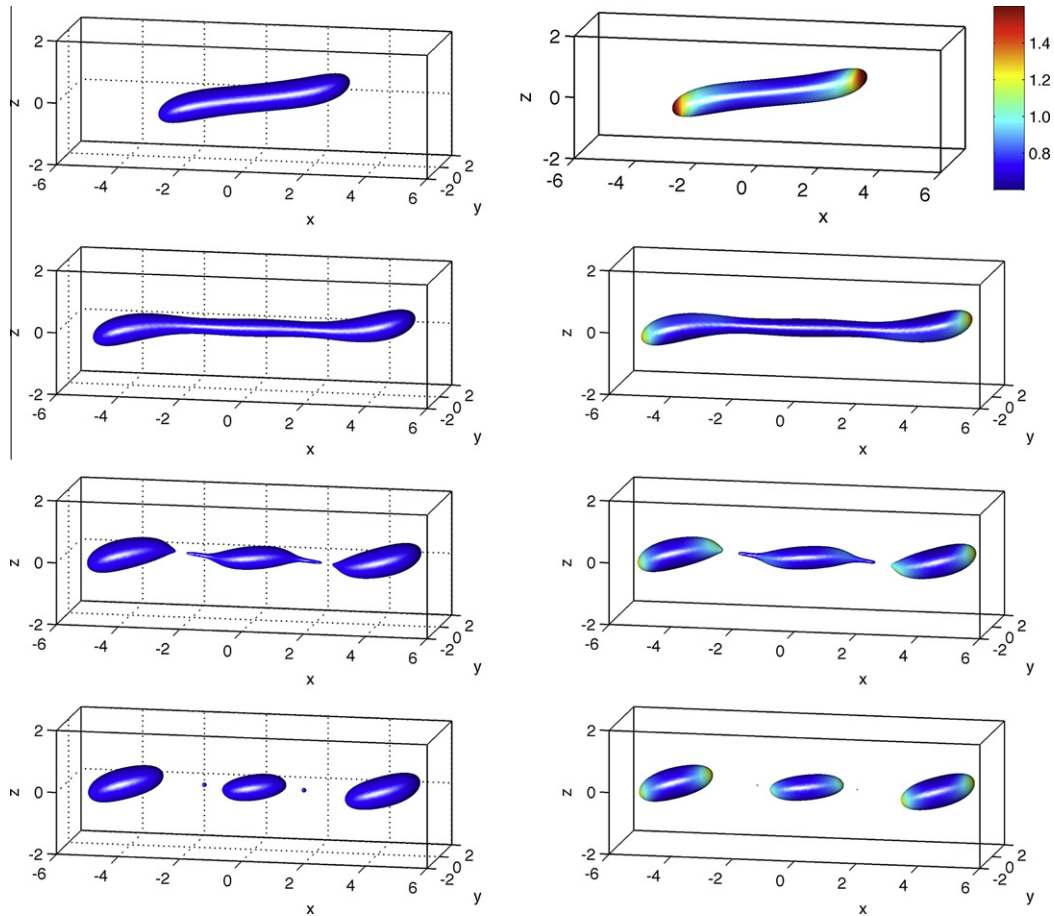


Fig. 15. Three-dimensional simulation of a drop in shear flow with interface thickness $\varepsilon = 1/80$. The left column shows a drop with no surfactant. The right column shows a drop with insoluble surfactant. The times shown are from top to bottom: 10, 25, 34, and 35.16.

drop sides is approximately 0.14 while at the drop tips the boundary layer thickness is approximately 0.69. For reference, the thickness, $O(\varepsilon)$, of the diffuse interface describing the drop boundary is approximately 0.06.

5.2.3. Convergence study

A convergence study was carried out for the fluid velocity, the interface surfactant and the bulk surfactant for the case where $Pe_F = 1$ and $Bi = 1$. Since the analytical solution is not available, we use the solution for a fine mesh as a reference solution and measure the deviation from this solution in the L_2 norm. The L_2 norm is calculated as

$$\|f\|_2 = \sqrt{\sum f^2}, \tag{57}$$

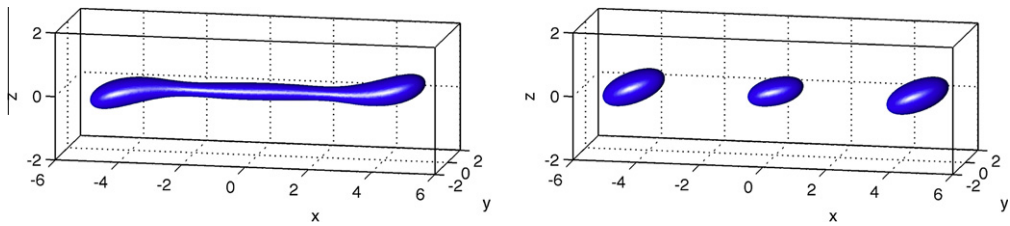


Fig. 16. Drop evolution in shear flow without surfactant at times $t = 25$ (left) and 35.16 (right) with $\varepsilon = 1/40$. All other parameters are the same as in Fig. 15.

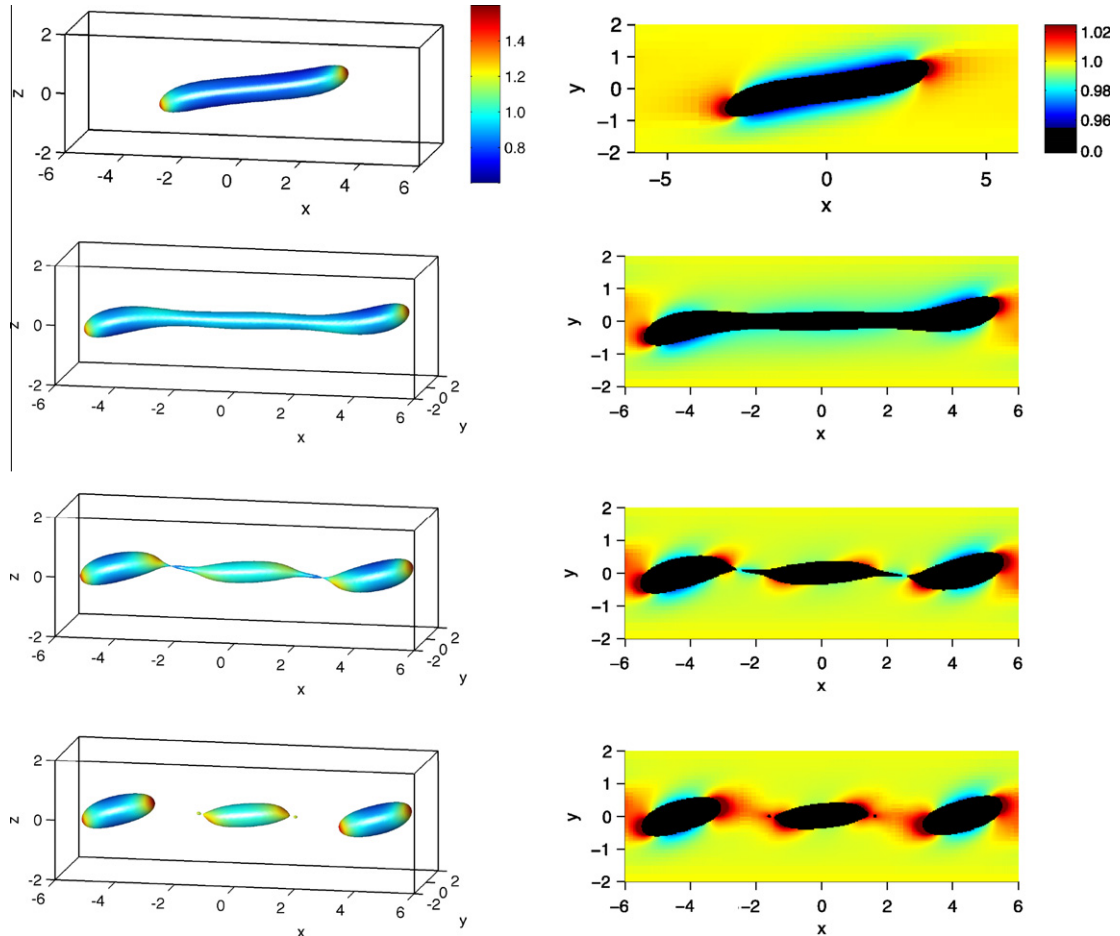


Fig. 17. Three-dimensional simulation of a drop in shear flow with soluble surfactant, $\varepsilon = 1/80$. The left column shows the drop shape with the surface colored according to the interface surfactant concentration. The right column shows a slice of the bulk surfactant concentration along the x -axis. The times shown are from top to bottom: 10, 25, 34 and 35.16.

and the rate of convergence as

$$\text{rate} = \frac{\ln(e_{2h}/e_h)}{\ln 2}, \quad (58)$$

where h denotes grid spacing. All simulations here are carried out on a uniform mesh. Since the variables are collocated, they do not coincide when the mesh is refined. Linear interpolation is used to interpolate the values.

The results are summarized in Table 1. The rate of convergence is around two for the velocities and one for the surfactant concentrations. The result for the surfactant concentrations are in line with what was found in [61].

5.3. Three-dimensional results

Finally, we present some results from three-dimensional simulations. In particular, we investigate drop break up in a shear flow. The grid spacing at the root level is $h_{\max} = 1/4$, three levels of refinement are used such that $h_{\min} = 1/32$. To elu-

cidate the effects of the interface thickness parameter ε , we simulated two different cases: $\varepsilon = 1/80$ and $1/40$. There are approximately 5–6 grid points across the interface for smaller ε . These simulations took approximately 3 days to complete on one core of an Intel i7 processor.

We take $Ca = 0.42$ and $Re = 0.4$, which while smaller than used in 2D are still above the critical threshold for the existence of steady state solutions. This allows us to run the simulation until the drop breaks up to demonstrate the ability of our method to handle topological changes. A similar simulation was presented in [66], although their simulation used $Re = 0$. For the surfactant, we assume a dilute concentration of $x = 0.1$ and a relatively strong coupling between surfactant and surface tension with $\beta = 0.2$. For the parameters associated with solubility, we again choose initial conditions at equilibrium, $k = 1/9$ and $h = 0.9$. The bulk Peclet number is set to $Pe_F = 10$.

The results for a clean drop and a drop covered by an insoluble surfactant are shown in Fig. 15 (left and right columns respectively). After an initial regime of stretching, necks form near the two drop tips. These become thinner until the drop breaks up. In each case, two tiny satellite drops can be observed after the necks pinch off. In the surfactant case, the satellite drops are very small and are covered with nearly uniform surfactant. We see that the surfactant-covered drop breaks up at an earlier time. There are two principal reasons for this behavior. First, the non-uniform distribution of surfactant leads to larger deformation, as shown in the previous sections, which thins the drop more than the clean case. Second, in the necks, the surfactant concentration is low, which gives a larger surface tension (similar to the surface tension of a clean drop). Since the necks are thinner in the surfactant-laden drop this leads to faster breakup.

In Fig. 16, the break-up of a clean drop is shown for a larger value of interface thickness ε ($\varepsilon = 1/40$ compared to $\varepsilon = 1/80$ in Fig. 15). For the larger value of ε , the drops break up at an earlier time and satellites are not observed since they are not resolved by this value of ε . A similar behavior is observed in the presence of surfactant.

The results for soluble surfactant are shown in Fig. 17, using $\varepsilon = 1/80$. The right column, which shows a slice of the bulk concentration, indicates that surfactant is adsorbed to the surface at the middle of the drop, and desorbed at the tips. This leads to a more uniformly distributed surfactant concentration. Comparing Fig. 17 with Fig. 15 we can see the soluble-surfactant-covered drop breaks up at a later time than the clean drop and the drop with insoluble surfactant. The principle reason is the adsorption and desorption of surfactant. Since adsorption occurs over a larger surface area than desorption, which just occurs at the drop tips, the interfacial surfactant concentration becomes larger across the entire drop. This means that the deformation becomes larger and that there is significant concentration of surfactant near the drop necks. This slows the thinning of the necks since the surface tension is smaller. Thus, the drop breaks up at a *later* time than for the clean drop and the drop with insoluble surfactant. This shows that the influence of solubility can have an important influence on drop dynamics.

6. Conclusions and future work

A diffuse-interface method to simulate two-phase flows with soluble surfactants was presented. The method handles advection, diffusion and adsorption/desorption of surfactant, and is easy to implement using standard numerical techniques.

As an example of the applicability of the method, results were presented on the influence of solubility on a drop in shear flow. It was shown that solubility could have a considerable influence on the flow dynamics. Simulations in 3D were also performed, which showed that the influence of soluble surfactants is important for the breakup behavior of the drop.

Although not presented here, asymptotic and numerical evidence suggests that the convergence to the sharp interface system is first order in the interface thickness parameter ε [61]. It may be possible to gain second order accuracy in ε by explicitly removing the corresponding term in the asymptotic expansion as can be done in the context of solidification to enable simulations with arbitrary kinetic coefficients [80,81]. Also, more consistent projection methods for collocated grids were recently presented in [82]. These two improvements to the proposed method should be explored. In addition, as we have seen, when the bulk Peclet number is large, thin boundary layers develop near the deforming drop. Adaptive meshes can enable accurate simulations for a range of Peclet numbers but in some applications, the bulk Peclet number can reach $10^5 - 10^6$ (e.g., [83]) at which point it is infeasible to use adaptive meshes. Instead, following the approach recently developed by Booty and Siegel [23], a singular perturbation analysis should be incorporated in the numerical algorithm.

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