

Solid-State and *in situ* NMR Spectroscopy in Materials Science

Natalia Fulik, Felicitas Kolbe, Silvia Paasch, Julia Pallmann, Tobias Pietsch, Marcus Rauche, Jana Richter, and <u>Eike Brunner</u>\* Faculty of Chemistry and Food Chemistry, Technische Universität Dresden, 01062 Dresden, Saxony, GERMANY eike.brunner@tu-dresden.de, https://tu-dresden.de/mn/chemie/anc/anc1





Alexander von Humboldt Stiftung/Foundation



Solid-state NMR spectroscopy delivers structural information based on the exploitation of short-range interactions. It does not require a long-range (crystalline) order. Based on continuous methodical progresses, it has meanwhile found numerous applications for the characterization of various materials as it will be demonstrated in this contribution for selected examples.

Based on strongly distance-dependent interactions such as the magnetic dipole-dipole interaction, solid-state NMR spectroscopy allows not only bulk materials characterization, but also the selective characterization of interfaces. This is exploited by our group, e.g., to study biominerals<sup>[1,2]</sup> and other hybrid materials such as metal-organic frameworks (MOFs).<sup>[3]</sup>

*In situ* studies of host-guest interactions are particularly powerful to characterize surfaces, e.g., in porous materials such as MOFs. Adsorption/desorption isotherms can be followed by observing the signals of the adsorbed gases. NMR-derived parameters can then be correlated with volumetric adsorption/desorption isotherms. Gases like <sup>129</sup>Xe, <sup>13</sup>CO<sub>2</sub>, and <sup>13</sup>CH<sub>4</sub> were already used for our investigations. <sup>129</sup>Xe NMR spectroscopy for example allows studying adsorption-induced structural transitions since it provides characteristic, structure-sensitive parameters like the chemical shift, the chemical shift anisotropy, signal intensities, and relaxation times.<sup>[4,5]</sup>

Finally, we are using a combination of solid- and liquid-state NMR spectroscopy for reaction monitoring in ionic liquids.<sup>[6]</sup>

## Acknowledgement

Thanks are due to the Deutsche Forschungsgemeinschaft (FOR 2038, FOR 2433, SPP 1708, SPP 1928) for financial support.

## **References:**

- D. Wisser, S. I. Brückner, F. Wisser, G. Althoff-Ospelt, S. Kaskel, E. Brunner, Solid State Nucl. Magn. Reson. 2015, 66/67, 33-39.
- (2) A. Jantschke, E. Koers, D. Mance, M. Weingarth, E. Brunner, M. Baldus, *Angew. Chem. Int. Ed.* **2015**, *54*, 15069-15073.
- (3) N. Kavoosi, V. Bon, I. Senkovska, S. Krause, C. Atzori, F. Bonino, J. Pallmann, S. Paasch, E. Brunner, S. Kaskel, *Dalton Trans.* 2017, 49, 4685-4695.
- (4) H.C. Hoffmann, B. Assfour, F. Epperlein, N. Klein, S. Paasch, I. Senkovska, S. Kaskel, G. Seifert, E. Brunner, J. Am. Chem. Soc. 2011, 133, 8681-8690.
- (5) J. Schaber, S. Krause, S. Paasch, I. Senkovska, V. Bon, D. Toebbens, D. Wallacher, S. Kaskel, E. Brunner, *J. Phys. Chem. C* 2017, *121*, 5195–5200.
- (6) M.F. Groh, S. Paasch, A. Weiz, M. Ruck, E. Brunner, Eur. J. Inorg. Chem. 2015, 24, 3991-3994.

## Second Interdisciplinary and Research Alumni Symposium iJaDe2018